

Photocatalytic Degradation of Methylene Blue in Aqueous Solution over Fe - doped and S - doped TiO₂ under Visible Light Ir

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ABSTRACT

The objectives of this work are to prepare Fe-doped and S-doped TiO₂ photocatalysts using the sol-gel method, and to investigate the photo-degradation of methylene blue (MB) in aqueous solutions under visible light irradiation. The preparation conditions, including the type and amount of dopants and calcinations temperature were studied the characteristics of the resulting photocatalysts were analyzed. Additionally, the optimum experimental conditions (including pH value and initial dye concentration) were evaluated.

The physical-chemical characteristics of the prepared photocatalysts were analyzed by BET, SEM, EDX, XRD, and UV-Vis. Results showed that the specific surface area of the photocatalysts is about 38.7~79.4 m²/g. XRD patterns indicated that the major crystalline type of the prepared TiO₂ is anatase. UV-Vis diffuse reflectance spectra showed that absorption of Fe-doped and S-doped TiO₂ in the visible light region was strengthened and the phenomena of red-shift was apparent. As for the photocatalytic degradation of MB, the pseudo-first-order rate equation can be used to fit the concentration vs. time data. It also revealed that the prepared photocatalysts had greater reaction rates and could react with MB more effectively under visible light irradiation. The S-doped TiO₂, using 0.15 mole thiourea as the dopant, and at the calcination temperature of 500 °C would exhibit the highest catalytic activity with respect to MB in this study. The optimum experimental conditions are found to be pH 3, and at the initial MB concentration of 10ppm. The removal efficiency of MB reached 88% after 60 minutes of visible light irradiation.

Keywords : sol-gel method、 photocatalysts、 visible light、 Fe、 S、 TiO₂、 MB

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