

Decomposition of indoor gaseous toluene using photocatalytic reactor

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ABSTRACT

In the present study, indoor level of gaseous toluene was decomposed by TiO₂-based visible light photocatalysts among which TiO₂ thin film was synthesized via a sol-gel method by using Titanium tetraisopropoxide (TTIP) as the precursor following by dip-coating process. In the Phase 1 study, the experiments were conducted in a bench scale glass cylindrical reactor illuminated by 300-400 nm UV light. The results indicated that low concentration of gaseous toluene was decomposed significantly by using TiO₂ thin film prepared from both Degussa P25 and TTIP. As the initial toluene concentrations were varied from 0.5 ppm to 6 ppm, the overall decomposition rate for UV-illuminated Degussa P25 was decreased approximately from 88% to 8%. As a comparison, sol-gel TiO₂ films prepared from TTIP showed more efficient photocatalytic ability under extremely low concentration at which 2 ppm toluene were decomposed completely at the gaseous flow rate of 200 mL min⁻¹. In the Phase 2 study, dye-sensitized TiO₂ photocatalysts were tested under visible light illumination. The experimental results showed that dye-sensitized TiO₂ photocatalysts prepared from TTIP can work under visible light illumination but its removal efficiency was not high. It was approximately 35% for toluene at 0.5 ppm. The results also indicated that dye-sensitized P25 with acid treatment can decompose low concentration of gaseous toluene and its removal efficiency was approximately 48% for toluene at 0.5 ppm but dye-sensitized P25 without acid treatment only works under UV-illuminated conditions. The commercial photocatalyst of Ag/TiO₂ was also tested but the results indicated that it almost can not decompose gaseous toluene at low concentration under visible light illumination.

Keywords : indoor air; photocatalytic; dye-sensitized TiO₂; toluene; visible light

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