

燃燒合成具多穩定相之金屬矽化物與硼化物

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摘要

本研究係以自持傳遞高溫合成法(Self-propagating High-temperature Synthesis, SHS), 在氬氣環境下進行燃燒合成之 鈮矽(Nb-Si)介金屬、鈮硼(Nb-B)化合物以及鉬矽(Mo-Si)介金屬三 大部份之實驗研究, 並於實驗中觀察各種不同莫爾比、試片密度 與預熱溫度對其火焰鋒面傳遞模式、燃燒溫度、火焰鋒面傳遞速度之影響, 並且詳細觀察燃燒合成產物與反應物元素當量比之關係。第一部份觀察各種不同莫爾比之鈮矽介金屬如: Nb₃Si、 Nb₅Si₃、 Nb₅Si₃與NbSi₂, 以及複合材料Nb₅Si₃/Nb組成特性。經由XRD分析其合成結果, 在Nb : Si = 5 : 3其主要生成產物為 Nb₅Si₃; Nb : Si = 3 : 2其主要生成產物為 Nb₅Si₃。而Nb : Si = 1 : 2除了生成NbSi₂介金屬, 尚殘留少許未反應之鈮與矽。於實驗中搭配Nb : Si = 3 : 1發現無法利用此方法引起自持燃燒傳遞, 因而不予以探討。而實驗結果顯示出Nb : Si = 5 : 3其燃燒溫度與火焰鋒面傳遞速度較其餘組態略高許多, 並且比較所有組態之燃燒溫度與火焰鋒面傳遞速度有相關一致性。在Nb₅Si₃/Nb複合材料方面, 將鈮粉從5 mol%增加至15 mol%其燃燒溫度與火焰鋒面傳遞速度都會隨著添加強化劑鈮粉的增加而降低, 並且經由XRD分析 其主要生成產物為 Nb₅Si₃以及少量未反應鈮粉殘留。根據組態5 : 3與1 : 2量測之燃燒溫度與火焰鋒面傳遞速度結果, 計算出Nb₅Si₃ 與NbSi₂介金屬反應活化能約為259.2 kJ/mol與160.8 kJ/mol。第二部份實驗為觀察各種不同莫爾比之鈮硼化合物組成特性, 如: Nb₃B₂、NbB、Nb₅B₆、Nb₃B₄以及NbB₂。經由XRD分析發現, 組態為1 : 1與1 : 2產物轉換率最佳, 皆可合成出NbB與NbB₂。而組態為3 : 2、5 : 6以及3 : 4則會同時生成兩者或兩者以上之 鈮硼化合物(Nb₃B₄、NbB、NbB₂), 並由實驗中發現無論搭配任何 組成皆無法合成出產物Nb₃B₂與Nb₅B₆。接著觀察Nb : B = 3 : 2燃燒溫度與火焰鋒面傳遞速度偏低的原因, 分析後發現是因為在燃燒合成的過程中有大部份未反應的鈮粉殘留。再者, 根據組態為1 : 1與1 : 2於預熱溫度200 °C以上, 火焰模式穩定平整的傳遞下, 量測之燃燒溫度與火焰鋒面傳遞速度結果, 計算出NbB與NbB₂反應活化能約為151.8 kJ/mol與132.4 kJ/mol。第三部份主要是觀察各種不同組態鉬矽介金屬, 其中包含Mo₃Si、Mo₅Si₃以及MoSi₂。產物合成結果在Mo : Si = 1 : 2 產物 轉換率最佳, 能夠完美的合成出MoSi₂ 介金屬。而Mo : Si = 3 : 1 與5 : 3 並無法利用此方法引起自持燃燒傳遞, 因而不予以探討。於實驗中搭配Mo : Si = 2 : 3與1 : 1 觀察其燃燒合成產物與反應物元素當量比之關係, 其同時生成產物MoSi₂ 與Mo₅Si₃ 介金屬以及未反應的鉬粉殘留。實驗結果顯示出Mo : Si = 3 : 2 與1 : 2 火焰鋒面傳遞速度較Mo : Si = 1 : 1 略快, 其燃燒溫度也較高。根據組態1 : 2 於實驗中量測之燃燒溫度與火焰鋒面傳遞速度結果, 計算出MoSi₂ 介金屬反應活化能約為34.9 kJ/mol。

關鍵詞: 鈮矽介金屬, 鈮硼化合物, 鉬矽介金屬, 自持傳遞高溫合成, 活化能, 預熱溫度, X 光粉末繞射分析儀

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