

# 應用模流分析技術於圓鋸機之設計及實驗

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## 摘要

本實驗研究是以自持傳遞高溫合成法 (Self-propagating High temperature Synthesis, SHS) 探討在不同之反應模式下, 所合成之金屬氮化物與碳化物。由於反應物反應形式之不同, 所得之產物與結果則有所差異, 因此本實驗中所探討之反應模式, 乃以反應物進行反應之形式區分, 主要為固相-氣相之燃燒反應、固相-固相之燃燒反應, 以及同時進行固相-固相與固相-氣相之燃燒反應, 並於此三種反應模式中特別針對其火焰鋒面傳遞模式、火焰鋒面傳遞速度及燃燒溫度變化等燃燒特性加以觀察, 並於固相-氣相反應模式中研究試片密度、預熱溫度、稀釋劑含量及氮氣壓力對於火焰傳遞速度與產物轉換率之影響; 在固相-固相反應模式中則探討試片密度、預熱溫度、稀釋劑含量對於產物組成之影響; 而同時進行固相-固相與固相-氣相反應模式中, 則是改變金屬粉末與碳粉末之混合比以合成不同 $[C]/([C]+[N])$ 函數比之金屬碳氮化物, 最後再將實驗所得之產物進行產物顯微結構之觀察與成份分析。在固相-氣相之燃燒反應中, 分別以鈮及鋁為反應物, 於0.274 ~ 4.238MPa之氮氣環境下燃燒合成金屬氮化物, 由於鈮之熔點甚高, 因此利用低熔點之鋁作為反應物, 以探討反應物熔點對於固相-氣相燃燒反應之影響, 實驗結果顯示, 在固相-氣相燃燒反應過程中, 均有出現二次燃燒現象, 而由於鋁之熔點較低, 因此在反應過程中明顯熔化變形, 由於反應需外部氮氣滲透參與反應, 因此試片密度與氮氣壓力則為固相-氣相反應中之重要參數, 降低試片密度與提高氮氣壓力均可有效提升產物之氮化率, 而氮-鋁反應時易因高溫熔化, 故需添加稀釋劑來加以改善, 當稀釋劑含量為50wt%時, 則可有效防止試片熔化而使產物氮化率提升。而固相-固相之反應模式則是將鈮與碳於氫氣中形成碳化鈮, 實驗結果顯示可合成出TaC與Ta<sub>2</sub>C兩種產物, 藉由熱電偶所量測之反應溫度約介於1700 ~ 1800°C之間, 而產物TaC外觀上則明顯膨脹且有明顯裂痕, 經成分分析後可知產物中會有少許鈮殘留, 而產物Ta<sub>2</sub>C中則有生成少許TaC, 只需提高產物密度即可改善, 而藉由火焰鋒面傳遞速度與反應溫度, 可計算出碳-鈮反應之活化能分別為TaC: 187.42及Ta<sub>2</sub>C: 298.97 kJ/mole。而將鈮與碳置於氮氣中燃燒, 即可同時進行固相-固相與固相-氣相反應而生成碳氮化鈮, 而碳含量與氮氣壓力則為影響反應之最主要參數, 增加氮氣壓力則可使產物氮化率提升, 含碳量越高時越不容易發生二次燃燒, 而產物分析中發現, 所得之產物均有鈮殘留, 且含碳量較低之產物易有中間產物Ta<sub>2</sub>N生成。

關鍵詞: 自持傳遞高溫合成法; 反應模式; 固相-氣相之燃燒反應; 固相-固相之燃燒反應; 同時進行固相-固相與固相-氣相之燃燒反應; 反應物熔點

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