

# 廢PU泡綿再利用研究

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## 摘要

聚氨基甲酸乙酯 (PU) 的使用與日俱增, 經使用後的處理與處置問題, 愈來愈受重視。廢PU的主要處理方法有掩埋處理法、資源回收法、物質回收法及化學處理回收法等。而就化學處理回收法來說, 廢PU泡綿經由添加化學物質, 在適當的觸媒催化及反應條件下產生化學反應, 反應後可得到PU的原料或初級石化原料。醇解反應 (GLYCOLYSIS) 為化學處理回收法之一種, 影響醇解反應的因素有溶劑的種類及濃度、觸媒配方及反應條件的控制。有鑑於此, 本研究探討上述因素對PU醇解過程及產物性質的影響, 以做為實廠設計之參考資料。本研究採用的樣品為軟質PU泡綿, 以添加不同配比的化學反應劑及觸媒, 在常壓及恆溫下進行醇解反應。實驗使用的反應劑為二甘醇 (DEG), 觸媒為醋酸鉀 (CH<sub>3</sub>COOK), 反應溫度為220。醇解產物性質分析項目包括氫氧基值、重量平均分子量、黏度及PU泡綿中-NCOO-官能基的轉化率。研究結果顯示, 以DEG / PU = 150%, KAC / PU = 1%, 反應時間90 MIN為適當反應配比、觸媒濃度及反應時間。在純化研究中發現, 第二階段蒸餾 (氣相溫度245~260) 之餾出物比例最高, 其氫氧基值與DEG者接近。從DEG添加量、-NCOO-轉化率、觸媒濃度及反應時間的分析結果, 得到適當反應配比及觸媒濃度 (DEG / PU = 150%, KAC / PU = 1%) 下之反應動力式可表示為:  $DX/DT=0.014 \times (1-X)^{3.71} \times (KAC)^{0.6} \times (DEG)^{1.12}$ , 其決斷係數為0.8202, 表示反應動力式是可接受的。

關鍵詞: 聚氨基甲酸乙酯、醇解反應、產物分析、反應動力、純化

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