

Studies on the Formation and Analysis of Antibacterial

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ABSTRACT

Although considered to be powerful veterinary medicines, diaveridine and sulfadiazine suffer from low water solubility that reduce their effectiveness. It is necessary to formulate these two drugs to enhance their water solubility and effectiveness. This study has focused on the development of an ideal formulation for these two drugs in hope to solve their solubility problem. In the context, several physical properties of these two drugs were reported, which includes thermal stability by DSC, UV absorbance, specific rotation, etc.. The analytical conditions for TLC, HPLC were also described. These two drugs were found stable at long exposure of light, heat and various pH. Nevertheless, the attempt to increase their solubility by addition of surfactant or salt formation were not yet successful. The ideal formation for two drugs is still to be seen. A method for simultaneous analysis of acetaminophen, aspirin, aluminum, ethoxybenzamide, codeine phosphate, thiamine disulfide, potassium guaiacol sulfonate, caffeine anhydrous, chlorpheniramine maleate, the common ingredients present in cold medicine was described. The method was based on reversed-phase ion-pairing high-pressure liquid chromatography. The best separation was achieved on a Vercopak C-18 column with a gradient elution, using a solvent system containing an acetate buffer (1:1, sodium pentane sulfonate (PICB-5)/sodium 1-heptanesulfonate (PICB-7), 5mM, with acetic acid to pH 3), and the methanol as mobile phase at a flow rate 1 ml/min, and a detection fixed wavelength of 280 nm. The method gave excellent results for three commercial products and a synthetic mixture containing eight major ingredients. The effects of the counter-ion concentration and the pH on the separation were also discussed.

Keywords : formulation ; analysis ; antibacterial

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