

# 具純銀導電性之微奈米級銀銅殼核粉末合成研究 = A Study on Synthesis of Micro/Nano Silver-Copper Core-Shell Powder with an

彭御賢、李清華

E-mail: 360088@mail.dyu.edu.tw

## 摘要

本研究主要分別探討以「銀銅共還原法」與「銅粉無電電鍍銀法」來合成銀銅殼核粉(亦即銅粉表面包覆薄銀層)，而本研究希望藉由此兩種合成方式，合成一中位粒徑小於 $10 \mu\text{m}$ 且具純銀導電性之微奈米銀銅殼核粉，以取代傳統成本較高之純銀粉，並提供電子業製作導電相關產品之用。

由「銀銅共還原法」之研究結果得知，當分別使用次亞磷酸鈉、抗壞血酸或異抗壞血酸鈉作為還原劑，以及控制不同之還原劑與銅銀金屬離子之莫耳比，並搭配於不同添加時間點下加入銀鹽進行粉體合成時，合成所得之粉體經由掃描式電子顯微鏡之背相散射顯像圖分析比對，顯示其不具有銀銅殼核之結構，另將該粉體製成導電塗層後經體電阻值分析後，發現該粉體不具導電性，故本研究使用之「銀銅共還原法」無法合成銀銅殼核粉。

另本研究所採用之「銅粉無電電鍍銀法」須透過兩階段進行合成銀銅殼核粉，第一階段係先透過田口法實驗設計，合成出可作為內層銅核使用之銅粉，本階段所得純銅粉之最佳合成參數包括將 $21.5 \text{ kg}$ 之硫酸銅與 $18.91 \text{ kg}$ 之磷酸鹽加入 $150 \text{ L}$ 之純水中，並控制反應溫度於 $70^\circ\text{C}$ 、攪拌轉速 $300 \text{ RPM}$ ，即可合成出粒徑為 $7.36 \mu\text{m}$ 之純銅粉。另第二階段則於此純銅粉表面上披覆上銀層，其最佳實驗參數為取 $2.53 \text{ g}$ 銅粉與 $0.08 \text{ g}$ 檸檬酸鹽加入 $100 \text{ ml}$ 純水中配置為銅粉分散液，另取 $1.17 \text{ g}$ 硫酸銀與 $0.065 \text{ g}$ 檸檬酸鹽配置成銀鹽溶液，於 $1000 \text{ RPM}$ 之轉速下，將此銀鹽溶液迅速倒入銅粉分散液中，待反應 $10$ 分鐘後，即可完成銅粉無電電鍍銀程序。此最佳合成粉體經掃描電子顯微鏡與表面元素分析儀分析後，証實該合成粉體為一具有奈米銀層之銀銅殼核結構，另該粉體所製作之導電塗層經檢測，其體電阻值與純銀粉體所製作之導電塗層完全相同( $1.43 \times 10^{-4} \Omega \cdot \text{cm}$ )，且該粉體經熱重分析儀分析後，可發現該粉體之氧化速率( $0.1465 \text{ Wt\%}/\text{min}$ )相較純銅粉明顯趨緩( $0.5745 \text{ Wt\%}/\text{min}$ )。

綜合上述，本研究所研發之「銅粉無電電鍍銀法」可合成一具有純銀導電性以及良好抗氧化性之微奈米銀銅殼核粉體，由於此粉體之生產成本遠較傳統純銀粉低廉，故適合廣泛推廣應用於導電材料之使用。

關鍵詞：共還原、無電電鍍、微奈米、田口法、銀、銅、導電、殼核

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