Treatment of pulp mill waste cooking liquor using an electrocoagulation method / 謝佳融 撰 .- 彰化縣大

謝佳融、彭元興,魏漣邦

E-mail: 354775@mail.dyu.edu.tw

ABSTRACT

In a kraft pulping process, effluents often entrain sulfides and mercaptans which give malodorous smell to the ambience. Therefore, an electrocoagulation system was used to treat the spent pulping liquor of a kraft pulp mill after adding sequestration chemicals such as calcium carbonate, iron sulfate, activated bentonite. Efficacies of the treatments in reducing sulfurous compounds in water were examined. After adding calcium carbonate and then applying an electrocoagulation treatment, the results showed good true color removal, and at a current density of 86.6 A/m2, hydraulic retention time (HRT) of 10 min, 85% of the colorant was removed. With regard to effluent COD, the aluminum electrodes outperformed cast iron electrodes, with respective removal rates of 50% and 40% at a current density of 86.6 A/m2 and HRT of 10 min. Removal of methanol was less than ideal, with a 44% removal rate under the optimal conditions. Removal of hydrogen sulfide reached ca. 80%, even the least performing aluminum electrode recorded a removal rate of 53%. Adding iron sulfate prior to electrocoagulation at 88.6 A/m2 and 10 min HRT produced result of maximal 72% true color removal. The COD removal rates were better, with 65% removal rate. Mediocre SS removal rate of 60~65% was obtained in this additive group. The methanol removal rate was the poorest, 30% at the best trial; removal of hydrogen sulfide, however attained 96% maximum, the best of 3 additives. The trials involving addition of activated bentonite prior to electrocoagulation at 86.6A/m2 and 10 min HRT produced a optimal true color removal rate of 87% with the aluminum electrodes which was not much different from the calcium carbonate group. The minimum removal rates 0f 49% for the group, however, was superior to the other 2 additives. COD removal reached 65% maximum for this group. Removal of SS was the best among the 3 groups, reaching 89% maximum. The same was observed for methanol removal rate of 46%. And hydrogen sulfide removal rate of 88% was deemed adequate. Overall, adding activated bentonite in conjunction with the electrocoagulation treatment produced superior results to those of the calcium carbonate and iron sulfite groups.

Keywords: Electrocoagulation, bentonite, pulping effluent, factorial design

Table of Contents

第一章 前言 1 1.1研究緣起 1 1.2研究動機 1 1.3研究目的 2 第二章 背景資料 3 2.1製漿廠廢水特性 3 2.2使用藥劑介紹 5 2.2.1 碳酸鈣 5 2.2.2硫酸鐵 5 2.2.3膨潤土 6 2.3電化學方法及優點 6 2.4電解膠凝法 7 2.5電解膠凝之應用 8 第三章 文獻回顧 10 第四章 實驗設計與方法 21 4.1實驗目的 21 4.2實驗設計 21 4.2.1操作參數及檢測項目 22 4.2.2階層設計 23 4.3實驗設備(電解膠凝系統) 24 4.4實驗步驟 28 4.5 檢測方法 29 4.6實驗設備 30 4.7實驗材料 31 4.8實驗藥品 31 第五章試驗結果與討論 32 5.1廢水水質分析 32 5.2處理水樣參數條件 33 5.3實驗結果 33 5.3.1 pH 34 5.3.1.1碳酸鈣組 34 5.3.1.2硫酸鐵組 36 5.3.1.3活化膨潤土組 38 5.3.2導電度 40 5.3.2.1碳酸鈣組 40 5.3.2.2硫酸鐵組 42 5.3.2.3活化膨潤土組 44 5.3.3真色色度 47 5.3.3.1碳酸鈣組 47 5.3.3.2硫酸鐵組 49 5.3.3.3活化膨潤土組 51 5.3.4化學需氧量(COD) 54 5.3.4.1碳酸鈣組 54 5.3.4.2硫酸鐵組 56 5.3.4.3活化膨潤土組 58 5.3.5水中懸浮固體物 61 5.3.5.1碳酸鈣組 61 5.3.5.2硫酸鐵組 63 5.3.5.3活化膨潤土組 65 5.3.6中醇 68 5.3.6.1碳酸鈣組 68 5.3.6.2硫酸鐵組 70 5.3.6.3活化膨潤土組 72 5.3.7硫化氫 74 5.3.7.1碳酸鈣組 75 5.3.7.2硫酸鐵組 77 5.3.7.3活化膨潤土組 79 5.3.8水中離子含量 81 第六章 結論與建議 84 6.1碳酸鈣組 84 6.2硫酸鐵組 84 6.3活化膨潤土組 85 6.4建議 86 參考文獻 87 附錄一 pH檢測結果 91 附錄二 導電度檢測結果 95 附錄三 真色色度檢測結果 99 附錄四 COD檢測結果 103 附錄五 SS檢測結果 107 附錄六 甲醇檢測結果 111 附錄七 硫化氫檢測結果 115

REFERENCES

1.Alinsafi A., Khemis M., Pons M.N., Leclerc J.P., Yaacoubi A., Benhammou A., Nejmeddine A. (2005). Electro-coagulation of reactive textile dyes and textile wastewater. Chemical Engineering and Processing 44:461-470. 2.Bektas N., Akbulut H., Inan H., Dimoglo A. (2004). Removal of phosphate from aqueous solutions by electro-coagulation. Journal of Hazardous Materials 106B: 101-105. 3.Bukhari A.A. (2007). Investigation of the electro-coagulation treatment process for the removal of total suspended solids and turbidity from municipal wastewater. Bioresource Technology. 99:914-921. 4.Garg A., Mishra I.M., Chand S. (2010). Effectiveness of coagulation and acid precipitation processes for the pre-treatment of diluted black liquor. Journal of Hazardous Materials 180:158-164. 5.Gao P., Chen X., Shen F., Chen G. (2005). Removal of

chromium (VI) from wastewater by combined electrocoaguiation-electroflotation without a filter. Sep Pur Technol 43:117-123. 6.Holt P., Barton G., Mitchell C. (1999). electrochemical as a wastewater treatment. The Third Annual Australian Environmental Engineering Research Event. 23-26 November Castlemaine, Victoria. 7.Inan H., Dimoglo A., Simsek H., Karpuzcu M. (2003). Olive oil mill wastewater treatment by means of electro-coagulation. Separation Purification Technology 36:23-31. 8. Wieckowska J. (1995). Catalytic and adsorptive desulphurization of gases. Catalysis Today 24:405-465. 9.Lai C.L., Lin S.H. (2003). Electro coagulation of chemical mechanical polishing (CMP) wastewater from semiconductor fabrication. Chemical Engineering Journal 95: 205-211. 10.Ma H., Wang B., Wang Y. (2007). Application of molybdenum and phosphate modified kaolin in electrochemical treatment of paper mill wastewater. Journal of Hazardous Materials 145:417-423. 11. Sridhar R., Sivakumar V., Prince Immanuel V., Prakash Maran J. (2011) Treatment of pulp and paper industry bleaching effluent by electro-coagulant process. Journal of Hazardous Materials 186:1495-1502. 12. Soloman P.A., Ahmed Bashab C., Velan M., Balasubramanian N., Marimuthu P. (2009). Augmentation of biodegradability of pulp and paper industry wastewater by electrochemical pre-treatment and optimization by RSM. Separation Purification Technology 69:109-117. 13. Yang C.L., Kravets G. (2000). Removal of chromium from abrasive blast media by leaching and electrochemical precipitation. Journal of the Air & Waste Management Association. 50(4): 536-542. 14. Xu X., Zhu X. (2004). Treatment of refectory oily wastewater by electro-coagulation process. Chemosphere 56:889-894. 15. Ugurlua M., Gurses A., Dogar C., Yalcm M. (2008). The removal of lignin and phenol from paper mill effluents by electro-coagulation. Journal of Environmental Management 87:420-428. 16. Zaroual Z., Azzi M., Saib N., Chainet E. (2005). Contribution to the study of electro-coagulation mechanism in basic textile effluent. Journal of Hazardous `Materials B131:73-78. 17.王文義(2001),利用電聚浮除法處理工業綜合廢水之研究,逢甲大學土木及水利工程研究所,碩士論文,台中。 18.孫晨光、高麗君、宋黑、丁迷理(2005),膨潤土在汙水處理中的應用與研究展望,河北工程學院。 19.紹紅、王冬梅、李穎惠、王恩 德(2004),改性膨潤土處理造紙廢水之研究,東北大學資源與土木工程學院,碩士論文,遼寧。 20.彭元興、王益真、余世宗、史濟元、 謝元昌、楊逸婷(2006),先驅廠級脈衝電凝系統在紙管用紙廠廢水回收在利用探討,第三十一屆廢水技術研討會:57,中華民國環境工程 學會,台中。 21.彭元興(2004),造紙產業用水管理,漿紙技術8(2):19-41。 22.彭元興、王益真、余世宗、史濟元、林逸汎、陳威存(2005) ,電氧化法應用在工業用紙廠廢水之探討,第三十屆廢水技術研討會:136,中華民國環境工程學會,中壢。 23.彭元興、王益真、史濟 元、張安毅、林逸汎(2004),工業用紙廠廢水回收再利用探討-先驅廠及脈衝電集系統的應用,清潔生產暨永續發展研討會,經濟部工 業局,台北。 24.張安毅(2005),脈衝電集法在工業用紙廠廢水之應用,碩士論文,大葉大學環境工程學系,彰化。 25.張志銘(2000),電 聚浮除配合逆滲透法處理石化廢水之研究,碩士論文,淡江大學水資源及環境工程學系,台北。 26.郭貴順(2006),以電聚浮除法處理化 妝品工業廢水,碩士論文,淡江大學水資源及環境工程學系,台北。 27.薛穆棨(2007),電聚浮除技術處理煉油廢水之實例探討,碩士論 文,國立中央大學環境工程研究所,桃園。 28.廖紋蘭(2005),石化工業廢水二級處理出流水再生利用技術之可行性研究,碩士論文,國 立成功大學環境工程系,台南。 29.鄭華安(2000),工業區廢水二級處理放流水回收再利用技術研究,碩士論文,國立成功大學環境工程 系,台南。 30.謝元昌(2007),電膠凝技術在製漿造紙廢水之應用研究,碩士論文,大葉大學環境工程學系,彰化。 31.蕭振宗(2010),電 膠凝技術應用特定造紙廢水處理之研究,碩士論文,大葉大學環境工程學系,彰化。