

Optimization of chemical precipitation of struvite for nutrient removal

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ABSTRACT

This study evaluated the efficiency of struvite precipitation from synthetic wastewater under the effects of various process parameters. The nutrient removal efficiency was investigated by analysis of struvite precipitation. With a synthetic wastewater, Box-Behnken design with the response surface methodology (RSM) was employed to determine the effects of pH value, temperature, Magnesium to Phosphorus molar ratio (Mg:P), mixing speed (rpm) and organic compounds on phosphorus recovery. Test results showed that PO₄-P recovery efficiency ranged from 56.04% to 90.04%. The results of ANOVA indicated that the proposed quadratic model predict the responses adequately. The experimental results for struvite precipitation first-order kinetics for the factors of mixing and organic contents showed that, a) mixing speed enhanced phosphorus removal efficiency, as seen when the mixing speed increased in range of experiment (50 to 100 rpm). The rate constants were found to be 0.813; 3.633 and 3.928 L.mMol-P⁻¹.min⁻¹ when the mixing speed are 50, 80, 100 rpm, respectively; b) organic substance (sugar) had little effect on the amount PO₄-P finally removed, but had significant effect the intrinsic rate constants. The rate constants are found to be 1.231; 3.023 and 1.49 L.mMol-P⁻¹.min⁻¹ for sugar concentration of 0.6; 1.02 and 1.05 g/L, respectively. The maximum NH₄-N recovery was 10.2% corresponded to the PO₄-P recovery of 90.04% and led to the conclusion that struvite precipitation was not the suitable method for Nitrogen recovery.

Keywords : Struvite、RSM、Box-Behnken design、ANOVA、first-order kinetic model.

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REFERENCES

- Ali, I. M., Schneider, P. A., & Hudson, N. (2005). Thermodynamics and solution chemistry of struvite. *Journal of Indian Institute of Science* , 85, 141-149.
- Ali, M. I. (2007). Struvite crystallization in fed-batch pilot scale and description of solution chemistry of struvite. *Chemical Engineering* , 85, 344 – 356.
- Ali, M. I., & Schneider, P. A. (2006). A fed-batch design approach of struvite system in controlled supersaturation. *Chemical Engineering Science* , 61, 3951-3961.
- Ali, M. I., & Schneider, P. A. (2008). An approach of estimating struvite growth kinetic incorporating thermodynamic and solution chemistry, kinetic and process description. *Chemical Engineering Science* , 63, 3514 -- 3525.
- Aslan, N., & Cebeci, Y. (2007). Application of Box – Behnken design and response surface methodology for modeling of some Turkish coals. *Fuel* , 86, 90 – 97.
- Atake, T. (2008). Application of calorimetry and thermodynamics to critical problems in materials science. *J. Chem. Thermodynamics* , 41, 1-10.
- Battistoni, B., De Angelis, A., Prisciandaro, M., Boccadoro, R., & Bolzonella, D. (2000). P removal from anaerobic supernatants by struvite crystallization: long term validation and process modelling. *Water Research* , 36, 1927-1938.
- Battistoni, P., Fava, G., Pavan, P., Musacco, A., & Cecchi, F. (1997).

Phosphate removal in anaerobic liquors by struvite crystallization without addition of chemicals: Preliminary results. *Water Research* , 31, 2925-2929. Bhuiyan, M. I., Mavinic, D. S., & Koch, F. A. (2007). Thermal decomposition of struvite and its phase transition. *Chemosphere* , 70, 1347-1356. Doyle, J. D., & Parsons, S. A. (2002). Struvite formation, control and recovery. *Water Research* , 36, 3925 – 3940. Doyle, J. D., Philp, R., Churchley, J., & Parsons, S. A. (2000). Analysis of struvite precipitation in real and sythetic liquors. *Trans IChemE* , 78, 480-488. El Diwani, G., El Rafie, S., El Ibiari, N. N., & El-Aila, H. I. (2006). Recovery of ammonia nitrogen from industrial wastewater treatment as struvite slow releasing fertilizer. *Desalination* , 214, 200 – 214. Etter, B., Tilley, E., Khadka, R., & Udert, K. M. (2010). Low-cost struvite production using source-separated urine in Nepal. *Water Research* , 45, 852-862. Ferreira, G. J., Bricker, S. B., & Simas, T. C. (2007). Application and sensitivity testing of a eutrophication assessment method on coastal systems in the United States and European Union. *Journal of Environmental Management* , 82, 433 – 445. Ferreira, S. L., Bruns, R. E., Ferreira, H. S., Matos, G. D., David, J. M., Brandao, G. C., et al. (2007). Box-Behnken design: An alternative for the optimization of analytical methods. *Analytica Chimica Acta* , 597, 179 – 186. Hong-Duck, R., & Sang-III, L. (2010). Application of struvite precipitation as a pretreatment in treating swine wastewater. *Process Biochemistry* , 45, 563 – 572. Huang, H., Xu, C., & Zhang, W. (2011). Removal of nutrients from piggery wastewater using struvite precipitation and pyrogenation technology. *Bioresource Technology* , 102, 2523 – 2528. Jaffer, Y., Clark, T. A., Pearce, P., & Parsons, S. A. (2001). Potential phosphorus recovery by struvite formation. *Water Research* , 36, 1834 – 1842. Kabdasli, I., Safak, A., & Tunay, O. (2008). Bench-scale evaluation of treatment schemes incorporating struvite precipitation for young landfill leachate. *Waste Management* , 28, 2386 – 2392. Kim, D., Kim, J., Ryu, H.-D., & Lee, S.-I. (2009). Effect of mixing on spontaneous struvite precipitation from semiconductor wastewater. *Bioresource Technology* , 100 (1), 74-78. Le Corre, K. S., Valsami-Jones, E., Hobbs, P., Jefferson, B., & Parsons, S. A. (2006). Agglomeration of struvite crystals. *Water Research* , 41, 419-425. Le Corre, K. S., Valsami-Jones, E., Hobbs, P., Jefferson, B., & Parsons, S. A. (2007). Struvite crystallisation and recovery using a stainless steel structure as a seed material. *Water Research* , 41, 2449-2456. Li, H., Zhao, G., Niu, S., & Luan, Y. (2007). Technologic parameter optimization of gas quenching process using response surface method. *Computational Materials Science* , 38, 561 – 570. Liu, Z., Zhao, Q., Lee, D.-J., & Yang, N. (2008). Enhancing phosphorus recovery by a new internal recycle seeding MAP reactor. *Bioresource Technology* , 99, 6488 – 6493. Morse, G. K., Brett, S. W., Guy, J. A., & Lester, J. N. (1997). Review: Phosphorus removal and recovery technologies. *The Science of the Total Environment* , 212, 69-81. Munch, E. V., & Barr, K. (2000). Controlled Struvite Crystallization for Removing Phosphorus from Anaerobic Digester Sidestreams. *Water Research* , 35, 151-159. Nelson, N. O., Mikkelsen, R. L., & Hesterberg, D. L. (2003). Struvite precipitation in anaerobic swine lagoon liquid: effect of pH and Mg:P ratio and determination of rate constant. *Bioresource Technology* , 89, 229 – 236. Nordstrom, F. L., & Rasmuson, A. C. (2008). Prediction of solubility curves and melting properties of organic and pharmaceutical compounds. *European journal of pharmaceutical sciences* , 36, 330 – 344. Nyenje, P. M., Foppen, J. W., Uhlenbrook, S., Kulabako, R., & Muwanga, A. (2010). Eutrophication and nutrient release in urban areas of sub-Saharan Africa — A review. *Science of the Total Environment* , 480, 447 – 455. Ohlinger, K. N., Young, T. M., & Schroeder, E. D. (2005). Kinetics and thermodynamics of struvite crystallization as it applies to phosphate recovery from municipal wastewater for agricultural fertilizer production. *Phosphorus Recovery* , 1-8. Pastor, L., Mangin, D., Ferrer, J., & Seco, A. (2010). Struvite formation from the supernatants of an anaerobic digestion pilot plant. *Bioresource Technology* , 101, 118 – 125. Quintana, M., Colmenarejo, M. F., Barrera, J., Sanchez, E., Garcia, G., Travieso, L., et al. (2008). Removal of phosphorus through struvite precipitation using a by-product of magnesium oxide production (BMP): Effect of the mode of BMP preparation. *Chemical Engineering Journal* , 136, 204 – 209. Quintana, M., Sanchez, E., Colmenarejo, M. F., Barrera, J., Garcia, G., & Borja, R. (2005). Kinetics of phosphorus removal and struvite formation by the utilization of by-product of magnesium oxide production. *Chemical Engineering Journal* , 111, 45 – 52. Rahaman, M. S., Ellis, N., & Mavinic, D. S. (2003). Effects of Various Process Parameters on Struvite Precipitation Kinetics and Subsequent Determination of Rate Constants. *Chemical Engineering* , 13, 535-542. Roncal-Herrero, T., & Oelkers, E. H. (2011). Experimental determination of struvite dissolution and precipitation rates as a function of pH. *Applied Geochemistry* , 3, 1-21. Ronteltap, M., Maurer, M., & Gujer, W. (2007). Struvite precipitation thermodynamics in source-separated urine. *Water Research* , 41, 977-984. Ronteltap, M., Maurer, M., Hausherr, R., & Gujer, W. (2009). Struvite precipitation from urine – Influencing factors on particle size. *Water Research* , 44, 2038-2046. Saidou, H., Korchef, A., Ben Moussa, S., & Ben Amor, M. (2008). Struvite precipitation by the dissolved CO₂ degasification technique: Impact of the airflow rate and pH. *Chemosphere* , 74, 338 – 343. Stratful, T., Scrimshaw, M. D., & Lester, J. N. (2001). Conditions influencing the precipitation of Magnesium Ammonium Phosphate. *Water Research* , 35, 4191 – 4199. Suzuki, K., Tanaka, Y., Kuroda, K., Hanajima, D., & Fukumoto, Y. (2005). Recovery of phosphorous from swine wastewater through crystallization. *Bioresource Technology* , 96, 1544 – 1550. Uludag-Demirer, S., Demirer, G. N., & Chen, S. (2005). Ammonia removal from anaerobically digested dairy manure by struvite precipitation. *Process Biochemistry* , 40, 3667 – 3674. Uysal, A., Yilmazel, Y. D., & Demirer, G. N. (2010). The determination of fertilizer quality of the formed struvite from effluent of a sewage sludge anaerobic digester. *Journal of Hazardous Materials* , 181, 248 – 254. Wang, J., Burken, J.G., and Zhang, X. (2006). Effect of seeding materials and mixing strength on struvite precipitation. *Water Research* , 78, 125-132. Wahab, M. A., Hassine, R. B., & Jellali, S. (2011). Removal of phosphorus from aqueous solution by *Posidonia oceanica* fibers using continuous stirring tank reactor. *Journal of Hazardous Materials* , 79, 1-9. Wu, D., Zhou, J., & Li, Y. (2009). Effect of the sulfidation process on the mechanical properties of a CoMoP/Al₂O₃ hydrotreating catalyst. *Chem. Eng. Sci.* , 64, 198-206. Ye, Z.-L., Chen, S.-H., Wang, S.-M., Lin, L.-F., Yan, Y.-J., Zhang, Z.-J., et al. (2009). Phosphorus recovery from synthetic swine wastewater by chemical precipitation using response surface methodology. *Journal of Hazardous Materials* , 176, 1083 – 1088. Yetilmezsoy, K., Demirel, S., & Vanderbei, R. J. (2009). Response surface modeling of Pb(II) removal from aqueous solution by *Pistacia vera* L.: Box – Behnken experimental design. *Journal of Hazardous Materials* , 171,

551 – 562. Alley, E. R. (2007). *Water Quality Control Handbook*. New York: The MacGraw-Hill Companies. Follett, R. F., & Hatfield, J. L. (2001). *Nitrogen in the Environment: Sources, Problems, and Management*. (R. F. Follett, & J. L. Hatfield, Eds.) Elsevier B. V. Haynes, W. M., & Lide, D. R. (Eds.). (2011). *CRC Handbook of Chemistry and Physics*. Chemical Databases Online (CHEMnetBASE). IPNI. (1999). *Better Crops*. Norcross: International Plant Nutrition Institute (IPNI). Khuri, A. I. (Ed.). (2006). *Response Surface Methodology and Related Topics*. Singapore: World Scientific Publishing Co. Pte. Ltd. Levenspiel, O. (1999). *Chemical Reaction Engineering (3rd Edition ed.)*. New Jersey: John Wiley & Sons, Inc. Montgomery, D. C. (2005). *Design and Analysis of Experiments*. Arizona, United States of America: John Wiley & Sons, Inc. Pelczar, J. M., Chan, E. C., & Krieg, N. R. (1993). *Microbiology: Concepts and Applications*. New York: McGraw-Hill, Inc. Rankins Jr, D. L. (1996). *Nutrient Requirements of Beef Cattle (7th Edition ed.)*. Washington, D.C, United States of America: The National Research Council. Reynolds, T. D., & Richards, P. A. (2009). *Unit Operations And Processes In Environmental Engineering*. Singapore: CENGAGE Learning. Stanton, T. L., & Whittier, J. (2010). *Urea and NPN for Cattle and Sheep*. Colorado State University. Colorado: Colorado State University. Yang, K., & El-Haik, B. S. (2009). *Design for Six Sigma: A Roadmap for Product Development*. New York: McGraw-Hill, Inc. Bishop, P. L. (2006). *Control of Struvite Deposition in Wastewater Treatment Plants*. 11th Annual Central States Water Environment Association Education Conference. Minnesota: Central States Water Environment Association. CEEP. (2003). *CEEP Report*. Bruxelles: Centre Europeen d ' Etudes sur les Polyphosphates (CEEP). CEEP. (2010). *CEEP Report*. Bruxelles: Centre Europeen d ' Etudes sur les Polyphosphates (CEEP). General-Chemical. (2003). *Struvite Control in Wastewater*. New Jersey: General Chemical. Imbrium-System. (2009). *Phosphorus/nitrogen pollution costs Americans \$4.3 billion each year*. Rockville, USA: Imbrium Systems. Mesner, N., & Geiger, J. (2010). *Understanding watershed*. Water Quality Extension, Utah State University. Stewart, W. M. (2002). *Commercial Phosphorus Fertilizer... Know Your Sources*. Potash & Phosphate Institute (PPI) and Potash & Phosphate Institute of Canada (PPIC). Norcross, Georgia 30092-2837 USA: Potash & Phosphate Institute (PPI) and Potash & Phosphate Institute of Canada (PPIC). Volterra, L., & Boualam, M. (2002). *Eutrophication and health*. Copenhagen: World Health Organization- European Commission. Wood, C. W., Mullins, G. L., & Hajek, B. F. (2002). *Phosphorus in Agriculture*. Soil Quality Institute, United States Department of Agriculture. Washington, DC, USA: National Cooperative Soil Survey. Barthelmy, D. (2009, December 31). *Mineralogy Database*. Retrieved February 2011, from: <http://www.webmineral.com/data/Struvite.shtml> Chandrasekaran, A. (2009). *Dr. A.Chandrasekaran's Homepage*. Retrieved February 2011, from: <http://chandrasekaran.tripod.com/uses.html> CIESE. (2008). *The Center for Innovation in Engineering and Science Education*. Johnson, L. (2003). (L. Johnson, Producer) Retrieved February 2011, from Spectrum Laboratories Inc Web site. LANTEC. (2006). *LANTEC Product, Inc Web site*. Retrieved February 2011, from: <http://www.lantecp.com/casestudy/cs49.pdf> Lenntech. (2009). *Lenntech Water treatment & purification Holding B.V Web site*. Retrieved February 2011, from: <http://www.lenntech.com/phosphorous-removal.htm> Mosaic, I. (2010). *back-to-basics*. Retrieved February 2011, from A Premier Soil Fertility Information Website: www.back-to-basics.net/stewardship/pdfs/animalenvironmentAn-4.pdf Ralph, J., & Chau, I. (2001). Retrieved February 2011, from Mineral Data Website: <http://www.mindat.org/min-3811.html> USCB. (2010). *United States Census Bureau*. Retrieved February 2011. USEPA. (1999). *United States Environmental Protection Agency*. Retrieved February 2011.